



پانزدهمین کنفرانس علمی ایران ۱۳-۱۴ شهریور ۱۳۹۰ - دانشگاه بوعلی سینا



A. Abbasi^a, M.A. Kiani^a, M.S. Rahmanifar^b, H.R. Ghenaatian^a, M.F. Mousavi^{a*}
^aDepartment of Chemistry, Tarbiat Modares University, P.O. Box 14115-175, Tehran, Iran
^bFaculty of basic science, Shahed University, Tehran, Iran
^{*}mfmousavi@yahoo.com, or mousavim@modares.ac.ir

Electrochemical supercapacitors possess much higher capacitance and specific energy than normal capacitors [1]. Recently, transition metal oxides have been great attention for using as electroactive materials in electrochemical supercapacitors due to their long cycle life, high specific capacitance, high conductivity and good electrochemical reversibility[2-3]. In recent reports, nanostructured electrode materials have attracted great interests since these electrodes show better rate capabilities than conventional electrodes composed of the same materials [4], and different techniques have been developed in order to prepare nanostructured transition metal oxides [5]. In this work electrochemical behaviors of NiMnO₃ nanosheets were studied. Nanosheets of NiMnO₃ were synthesized directly by molten-salt procedure. The morphology and size of the as-prepared NiMnO₃ samples were investigated using Scanning Electron Microscopy (SEM). The cyclic voltammetric and electrochemical impedance spectroscopy were carried out for electrochemical study. Galvanostatic charge-discharge experiments used in order to investigate the applicability of the system as a redox supercapacitor. The XRD clearly revealed the formation of NiMnO₃. The SEM images showed that NiMnO₃ nanosheets consisted of uniform sheets with thickness in the range of 40 to 70 nm. Based on the galvanostatic charge-discharge results, high specific capacitance of 175 F.g⁻¹ was obtained, at constant discharge current of 1 mA cm⁻².

Reference

- [1] X.M. Liu, X.G. Zhang, S.Y. Fu, Materials Research Bulletin, 41 (2006) 620-627.
- [2] Yuan, X.H. Xia, J.B. Wu, J.L. Yang, Y.B. Chen, S.Y. Guo, Electrochimica Acta, 56 (2011) 2627-2632.
- [3] A. Zolfaghari, F. Ataherian, M. Ghaemi, A. Gholami, Electrochimica Acta, 52 (2007) 2806-2814.
- [4] H.R. Ghenaatian, M.F. Mousavi, S.H. Kazemi, M. Shamsipur, Synthetic Metals, 159 (2009) 1717-1722.
- [5] M. A. Kiani, M. F. Mousavi, S. Ghasemi, Journal of Power Sources 195 (2010) 5794.